

Reinhold Environmental Ltd.



2008 NO_x-Combustion Round
Table & Expo Presentation

February 4-5, 2008 in Richmond, VA

VOSTEEN Consulting GmbH

Thermal Engineering and Environmental Protection

**Scale Up
of Bromine Based Mercury Oxidation
in Coal Combustion and Waste Incineration (Part B)**

Bernhard W. Vosteen, Vosteen Consulting*)
Mark S. Berry, Southern Company Services
Rico Kanefke, Currenta

*) email: info@vosteen-consulting.de
Cologne, Germany
phone: + 49 221 680 098 22

Reinhold Environmental
NOx-Combustion Round Table

Richmond VA, Thursday, February 5th, 2008

In the following, further two parts of this workshop:

What we are mainly speaking about?

Some drops of BRAVO to the coal
(Bromine Additive Vosteen)

+

Some drops of PRAVO to the WFGD
(Precipitation Agent Vosteen)

might solve your mercury problem

in the most cost competitive way

Vosteen Consulting owns a General License granting worldwide exclusive marketing rights



Bayer Industry Services

Bayer Industry Services GmbH & Co. OHG
(Leverkusen, GER)



CURRENTA 

Currenta GmbH & Co OHG
former BAYER Industry Services
4 Rotary Kiln Incinerators
for Hazardous Wastes

**Joachim Beyer, Theodor Bonkhofer,
Rico Kanefke**

VOSTEEN CONSULTING GMBH
Thermal Engineering and Environmental Protection



US006878358B2

Bromine enhanced mercury abatement Invented by Prof. Vosteen in 2001

German Patent DE 10 233 173
granted 2005

US Patent 6 878 358 *)
granted 2005

Canadian Patent 2 435 474 *)
granted 2006

European Patent 2 435 474
granted 2008

*) actual owner: Vosteen Consulting GmbH

Patent applications pending
in other countries

(12) **United States Patent**
Vosteen et al.

(10) Patent No.: **US 6,878,358 B2**
(45) Date of Patent: **Apr. 12, 2005**

(54) **PROCESS FOR REMOVING MERCURY
FROM FLUE GASES**

(56) **References Cited**
U.S. PATENT DOCUMENTS

(75) Inventors: **Bernhard Vosteen**, Cologne (DE);
Joachim Beyer, Kuerten (DE);
Theodore-Gerhard Bonkhofer, Essen
(DE); **Olaf Fleth**, Grevenbroich (DE);
Andrea Wieland, Maria Rojach (AT);
Andreas Pohontsch, Goerlitz (DE);
Rico Kanefke, Merseburg (DE); **Ewa**
Standau, Merseburg (DE); **Claus**
Mueller, Kuerten (DE); **Michael Nolte**,
Goslar (DE); **Heinz Koester**, Ingelheim
(DE)

5,238,488 A * 8/1993 Wilhelm 75/742
5,900,042 A * 5/1999 Mendelsohn et al. 75/742
2002/0114749 A1 8/2002 Cole
2002/0114750 A1 8/2002 Holste et al.

(73) Assignee: **Bayer Aktiengesellschaft**, Leverkusen
(DE)

FOREIGN PATENT DOCUMENTS
CA 2150529 12/1995
DE 42 18 672 C1 8/1993 F23G/7/00
DE 44 22 661 A1 1/1996 B01D/53/34
DE 19850054 A1 * 5/2000 B01D/53/77
EP 0 208 036 B1 10/1989 B01D/53/34
EP 0 709 128 A2 5/1996 B01D/53/94
JP 02000197811 A * 7/2000 B01D/53/64

(*) Notice: Subject to any disclaimer, the term of this
patent is extended or adjusted under 35
U.S.C. 154(b) by 24 days.

OTHER PUBLICATIONS
Shoichi, Takao et al; 2000/197811 A; Patent Abstracts of
Japan; Jul. 18, 2000.

(21) Appl. No.: **10/430,088**

(22) Filed: **May 6, 2003**

(65) **Prior Publication Data**

US 2004/0086439 A1 May 6, 2004

Related U.S. Application Data

(63) Continuation-in-part of application No. 10/202,571, filed on
Jul. 24, 2002.

(30) **Foreign Application Priority Data**

Jul. 22, 2002 (DE) 102 33 173

(51) **Int. Cl.⁷** **B01D 53/00**; C22B 43/00

(52) **U.S. Cl.** **423/210**; 75/670; 75/742

(58) **Field of Search** 423/210; 75/670,
75/742

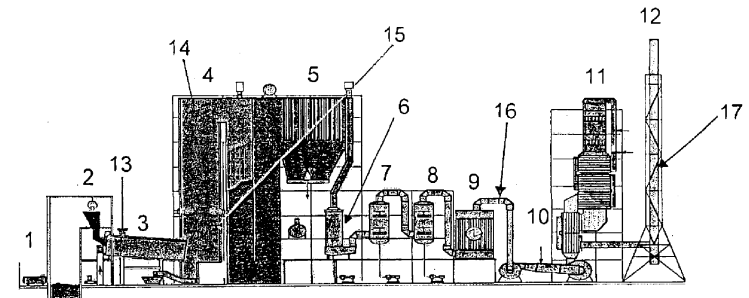
* cited by examiner

Primary Examiner—Stanely S. Silverman
Assistant Examiner—Jonas N. Strickland
(74) *Attorney, Agent, or Firm*—Norris McLaughlin &
Marcus PA

(57) **ABSTRACT**

Process for removing mercury from flue gases of high-
temperature plants, in particular power stations and waste
incineration plants in which a bromine compound is fed to
the multistage furnace and/or the flue gas in a plant section
downstream of the furnace, the temperature during contact
of the bromine compound with the flue gas being at least
500° C., preferably at least 800° C. The combustion is
carried out in the presence of a sulphur compound, in
particular sulphur dioxide. Subsequently to the furnace, the
flue gas is subjected to an optional multistage cleanup for
removing mercury from the flue gas, which cleanup com-
prises a wet scrubber and/or a dry cleanup.

16 Claims, 9 Drawing Sheets



German Licensees of Vosteen Consulting



Town Karlsruhe – Wastewater Treatment Plant
**2 Stationary Fluidized Bed Combustors
for Sewage Sludge**
applying NaBr and PRAVO® since 2007
Martin Maurer, Roland Milz

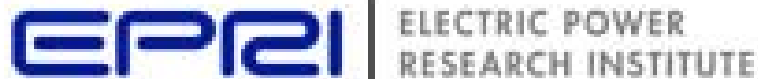


EGLV – Central Sludge Treatment Plant Bottrop
**2 Stationary Fluidized Bed Combustors
for Sewage Sludge**
applying NaBr since 2004
Testing PRAVO® in 2004 and again in April 2008
Falko Lehrmann, Günter Schwabe

US Licensees of Vosteen Consulting



ALSTOM-Environmental Control Systems,
Knoxville/TN, USA,
exclusive licensee for North America
KNX™ Coal Additives and Systems
as trademark of ALSTOM Power Inc.
**Michael J. Rini, John Buschmann,
Leif Lindau**



Electric Power Research Institute,
Palo Alto/CA, USA,
research grant
for demonstration test runs
e.g. at the sites Monticello, Plant Miller
George Offen, Ramsay Chang



Southern Company Services, Inc.
Birmingham/AL, USA,
specific licensee for SC-utilities only
Larry Monroe, Mark Berry, Nick Irvin 6

URS

Carl Richardson

Katherine Dombrowski

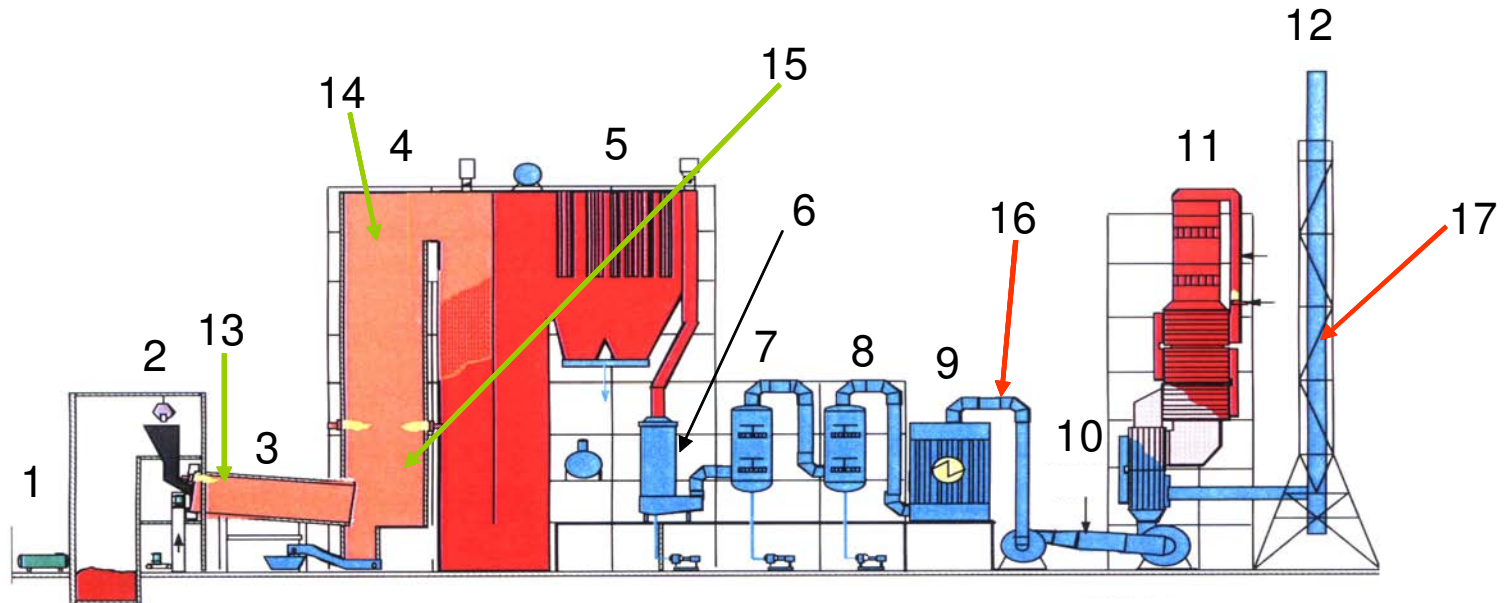
Jennifer Paradise

Gary Blythe

John Currie

et al.

1st Successful Industrial Application in operation since 2001 (4 Units)

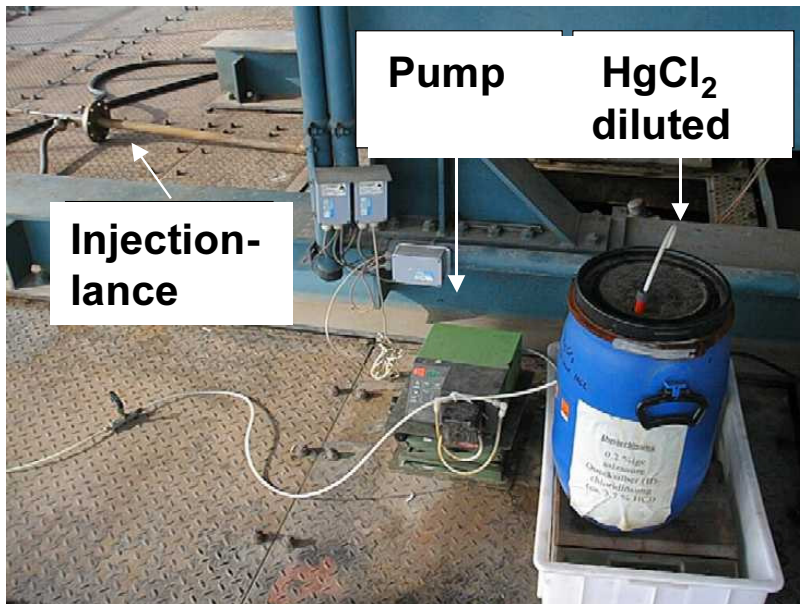


Hazardous Waste Combustion plants with 30,000. .. 80,000 Nm³/h dry each, CURRENTA GmbH & Co OHG in Leverkusen, Dormagen and Uerdingen, Germany. Waste heat recovery boiler (5), multistage scrubber (6, 7, 8), wet ESP (9), DeNOx-SCR (11)

Injection ports 13, 14 and 15 for continuous or discontinuous Hg-spikin
Measuring ports 16 and 17 with CEMs, partially used for mercury speciation 8

Hg-Spiking

Continuously



at top of after burning chamber

100 ... 10,000 ... 20,000 $\mu\text{g}/\text{Nm}^3$ dry

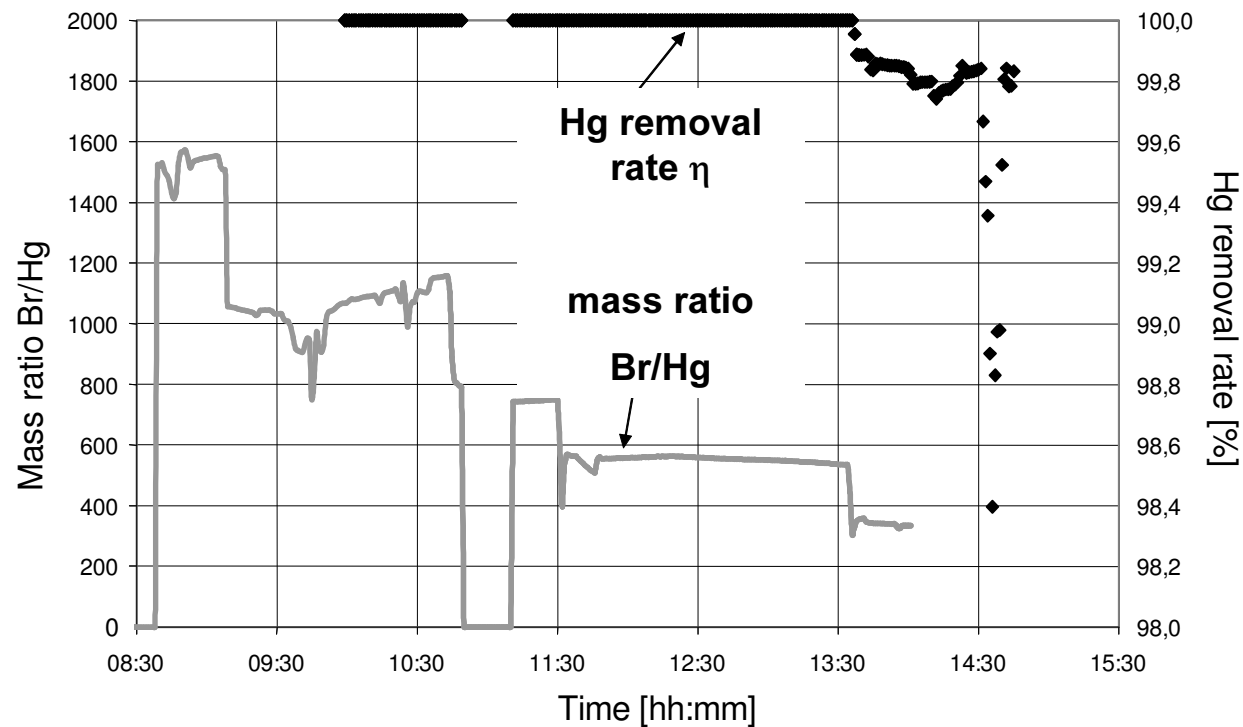
Discontinuously



at bottom of after burning chamber

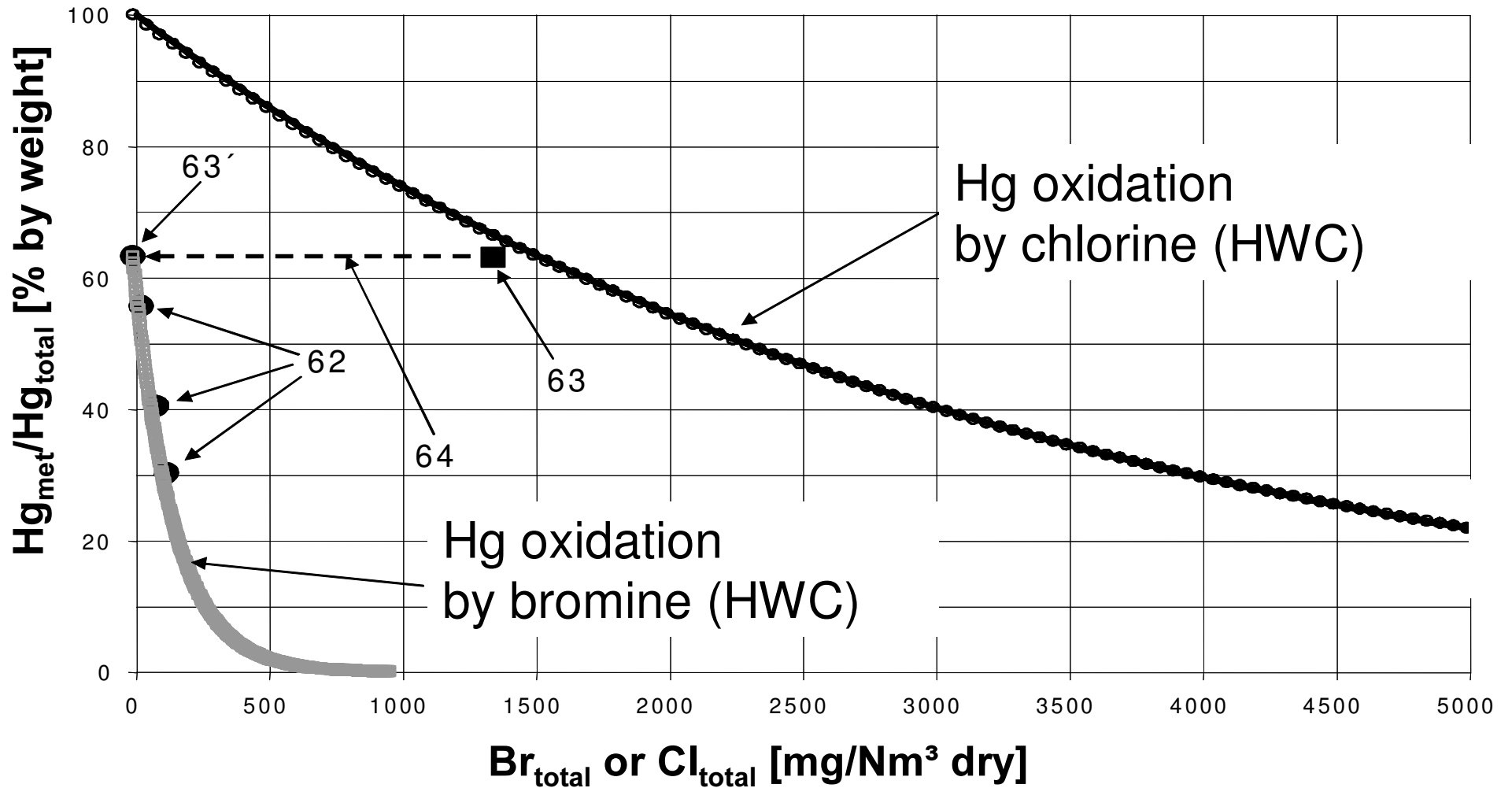
500,000 ... 80 Mio $\mu\text{g}/\text{Nm}^3$ dry
„Hg-bombs“ ---> „Hg-clouds“

**diploma thesis of Rico Kanefke, 2001,
high Hg_{total} in the flue gas ($9600 \mu\text{g Hg}/\text{Nm}^3$ dry)**



**Mass ratio for complete mercury removal:
 $\text{Br}/\text{Hg} = 500$ ("without SCR")**

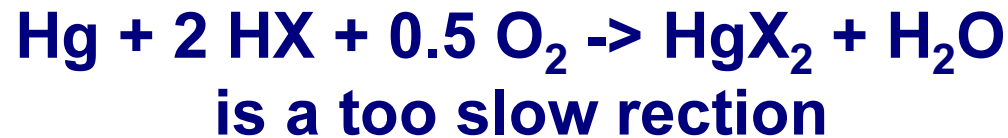
low Hg_{total} in the flue gas ($150 \mu g Hg/Nm^3$ dry)



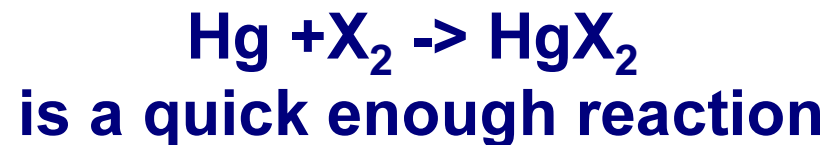
Why Bromine instead of Chlorine?

Working hypothesis

„Hg not reacting with HX“



„Hg reacting with X_2 or related radicals“



There is much more Br_2 than Cl_2 formed

Chlorine Enhanced Hg-Oxidation

Relevant global reactions



Chlorine less effective in Hg-oxidation

because $\text{Cl}_2/\text{Cl}_{\text{total}} \ll 1$

(e.g. < 4 % in HWC and < 1 % in coal combustion)

SO₂ is consuming Cl₂ during boiler passage

Bromine Enhanced Hg-Oxidation

Relevant global reactions



Bromine highly effective in Hg-oxidation

because $\text{Br}_2/\text{Br}_{\text{total}} \rightarrow 1$

SO_2 is not consuming Br_2 during boiler passage

(at temperatures $> 100 \text{ }^\circ\text{C}$)

Poster-Presentation Vosteen at Air Quality IV Conference, Arlington, September 2003



Vosteen Consulting GmbH
Thermal Engineering and Flue Gas Cleaning
info@vosteen-consulting.de



Mercury-Related Chemistry in Waste Incineration and Thermal Process Flue Gases

Bernhard W. Vosteen, Vosteen Consulting GmbH, Cologne (Germany)
Richard Ullrich, WastePro Engineering Inc., Kennett Square, PA

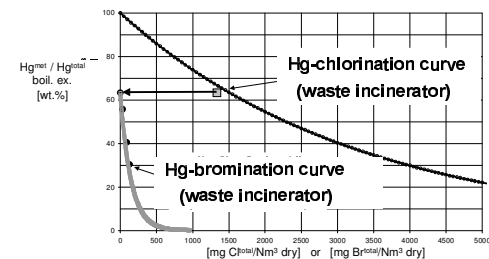
Bromine Enhanced Hg-Oxidation

Main global reactions	(hypothesis Vosteen)	Thermodynamics in boiler temperature range
$4 \text{HBr} + \text{O}_2 \leftrightarrow 2 \text{H}_2\text{O} + 2 \text{Br}_2$	Bromine-Deacon-Reaction	$\Delta_r G < 0$
$\text{SO}_2 + \text{Br}_2 + \text{H}_2\text{O} \leftrightarrow \text{SO}_3 + 2 \text{HBr}$	Bromine-Griffin-Reaction	$\Delta_r G \gg 0$
$\text{SO}_2 + \frac{1}{2} \text{O}_2 \leftrightarrow \text{SO}_3$	SO_2/SO_3 -Konversion	$\Delta_r G < 0$
$\text{Hg} + \text{Br}_2 \leftrightarrow \text{HgBr}_2$	direct Hg-Bromination ¹⁾	$\Delta_r G < 0$
$\text{SO}_2 + \text{Br}_2 \leftrightarrow \text{SO}_2\text{Br}_2$	Bromine-Sulfurylisation	$\Delta_r G \ll 0$
$\text{Hg} + \text{SO}_2\text{Br}_2 \leftrightarrow \text{HgBr}_2 + \text{SO}_2$	Hg-Bromination by SO_2Br_2	$\Delta_r G < 0$

¹⁾ $\text{Hg} + 2 \text{HBr} + \frac{1}{2} \text{O}_2 \leftrightarrow \text{HgBr}_2 + \text{H}_2\text{O}$ indirect Hg-Bromination in boiler not relevant,
but in SCR (V_2O_5 as oxygen donor)

**Bromine > 25 times more effective for Hg^{met} oxidation than chlorine
in waste incineration as well as in coal combustion**

(BAYER patent applications pending world wide)



Vosteen's 2003 comment on mercury bromination
**Bromine 25 times more effective than chlorine
for Hg^{met}-oxidation in waste incineration
as well as in coal combustion**

**Following this presentation at AQ IV
some bromine related activities
came up in the States as well in
industry as in science**

Internal US EPRI Report, March 2006

Chapter 3: „Boiler Chemical Additives“

Chemical Addition Tests Performed at Host Sites

Unit	Test Period	Reagents Tested	Addition Method
Laskin 2	May 2002*	NaCl, CaCl ₂ , Fe/NaCl	Solid addition to coal feed
	Sept. 2003*	MgCl ₂ , HCl	Liquid addition to boiler
	Aug. 2005*	CaCl ₂ , CaBr ₂	Solid addition to coal feed
Meramec 2	Sept. 2004**	KNX, SEA2	Liquid addition to coal feed
Baldwin 3	Aug. 2005*	CaCl ₂ , CaBr ₂	Liquid addition to boiler
Limestone 1	Sept.–Oct. 2004*	CaCl ₂ , CaBr ₂	Liquid addition to boiler
Monticello 3	Oct.–Dec. 2005**	CaCl ₂ , CaBr ₂	Liquid addition to coal feed
Milton R. Young 2	April 2005**	CaCl ₂ , MgCl ₂ , SEA2	Liquid addition to coal feed
Stanton 10	April 2002*	NaCl, MgCl ₂ , FeCl ₂	Liquid addition to boiler

* - EPRI-conducted tests

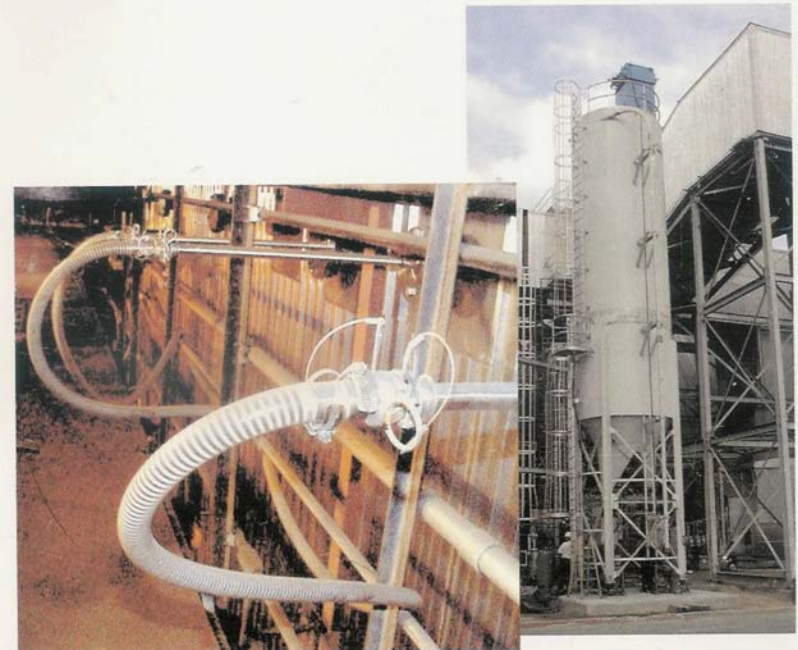
** - Tests conducted as part of DOE-NETL Cooperative Agreement Programs

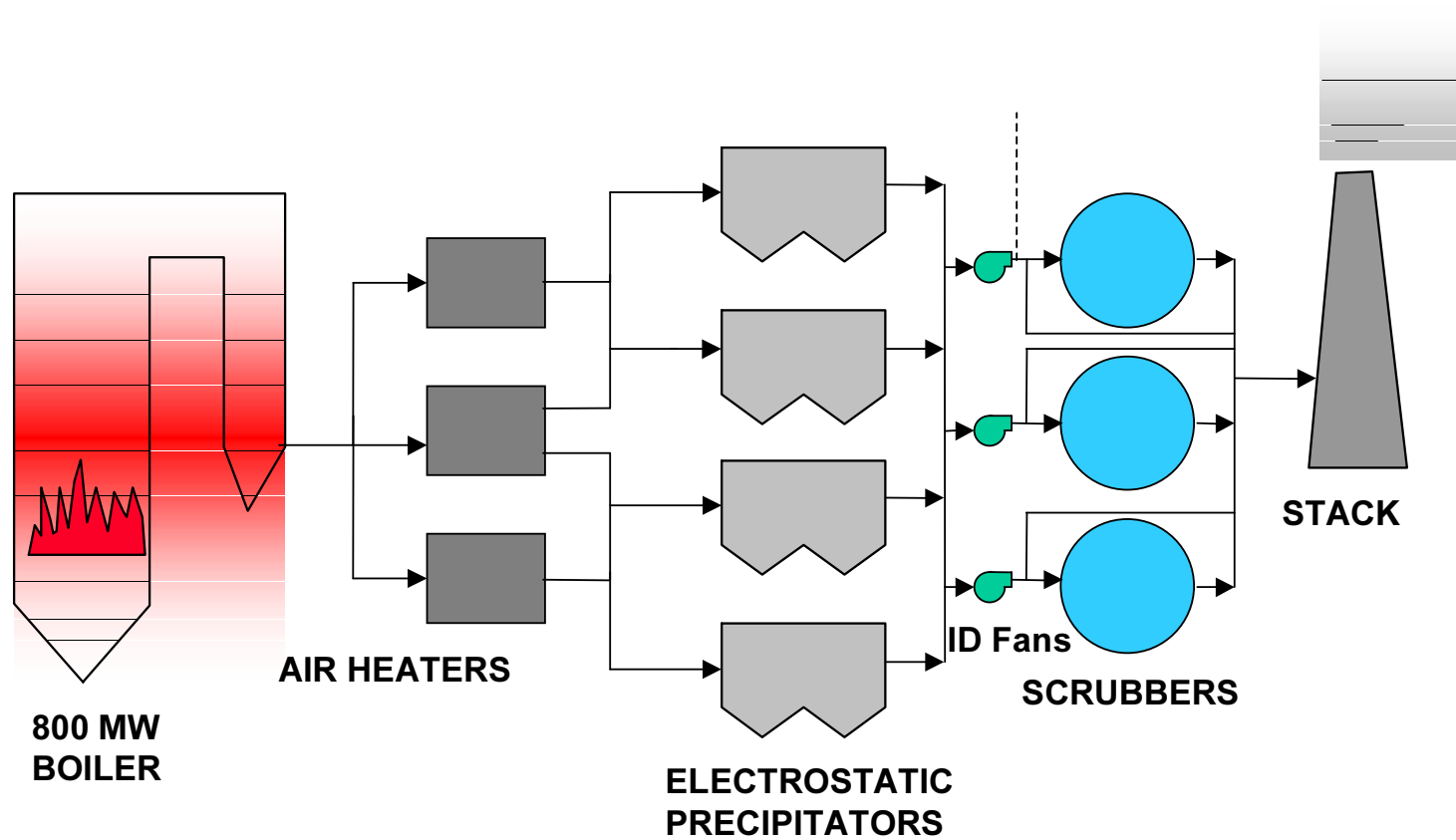
EPRI | ELECTRIC POWER RESEARCH INSTITUTE

Status of Mercury Control Technologies: Activated Carbon Injection and Boiler Chemical Additives



Technical Report

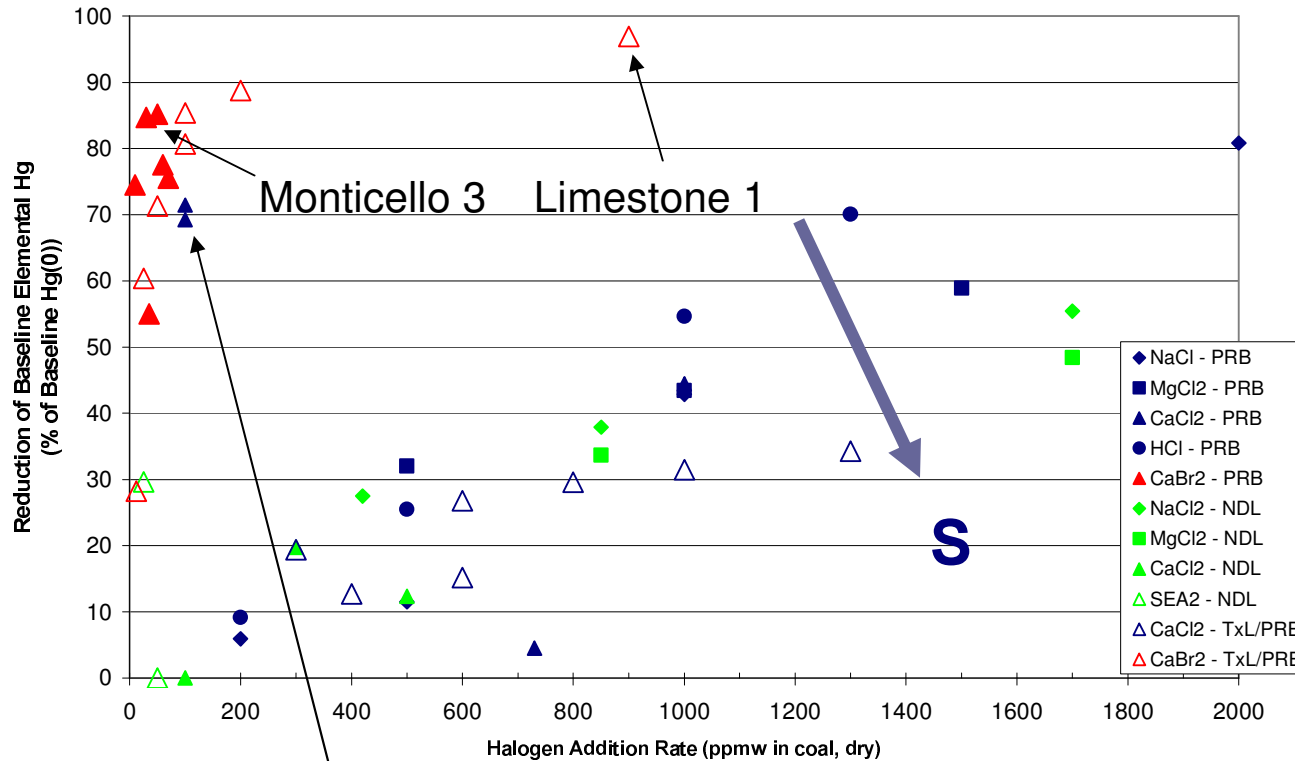




TXU Energy's Monticello Steam Electric Station Unit 3 (793 MWe)

Reduction of Baseline Elemental Mercury

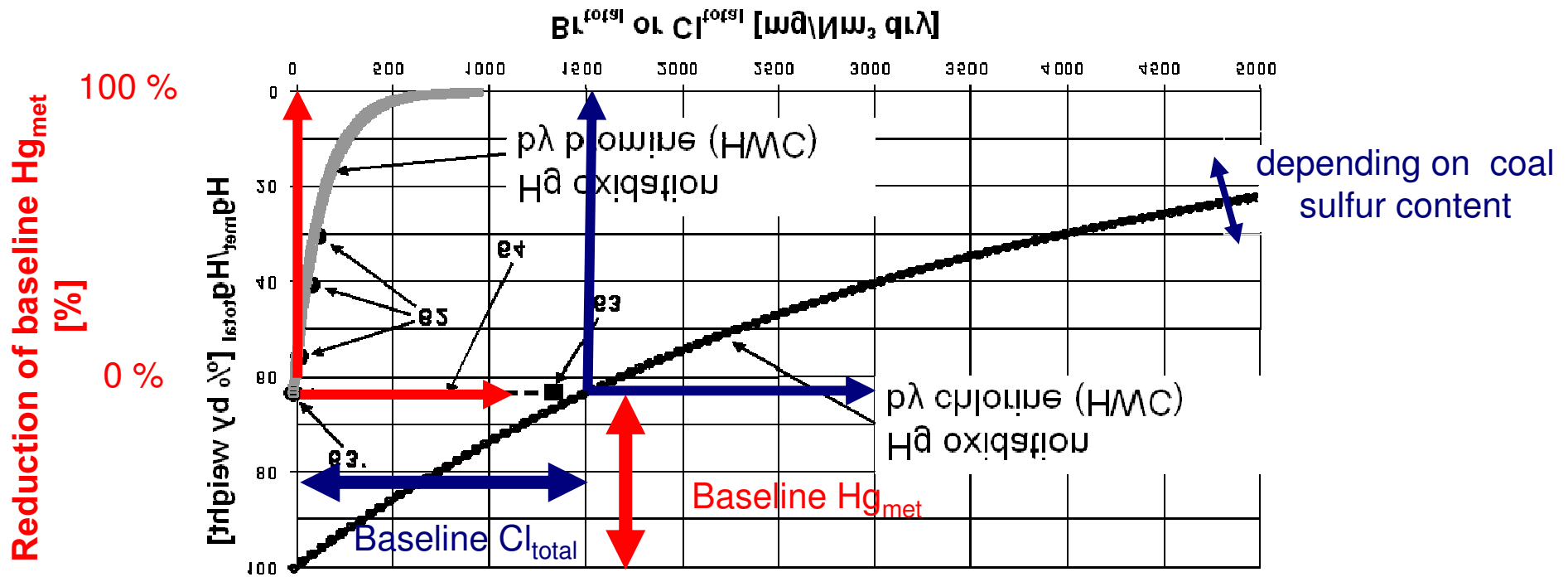
Overview over all sites tested by EPRI
 in 2002 ... 2005 (chlorine) and repectively 2004 ... 2005 (bromine)



"Halogen containing SEA2 (proprietary), composition not revealed"

bromination

chlorination



**Halogen addition
[mg/Nm³ dry]**

Comparing Waste combustion with EPRI test runs at PC-fired power stations in 2001 ... 2005 (chlorination) and 2004 ... 2005 (bromination)

Lower mercury concentration in coal combustion (factor 85/15 = factor 6) and 6.5 Nm³ dry/kg PRB ->

The comparison shows that the results attained at the HWC are not only qualitatively the same, but quantitatively too!

University R&D:

VOSTEEN Consulting GmbH

Thermal Engineering and Environmental Protection

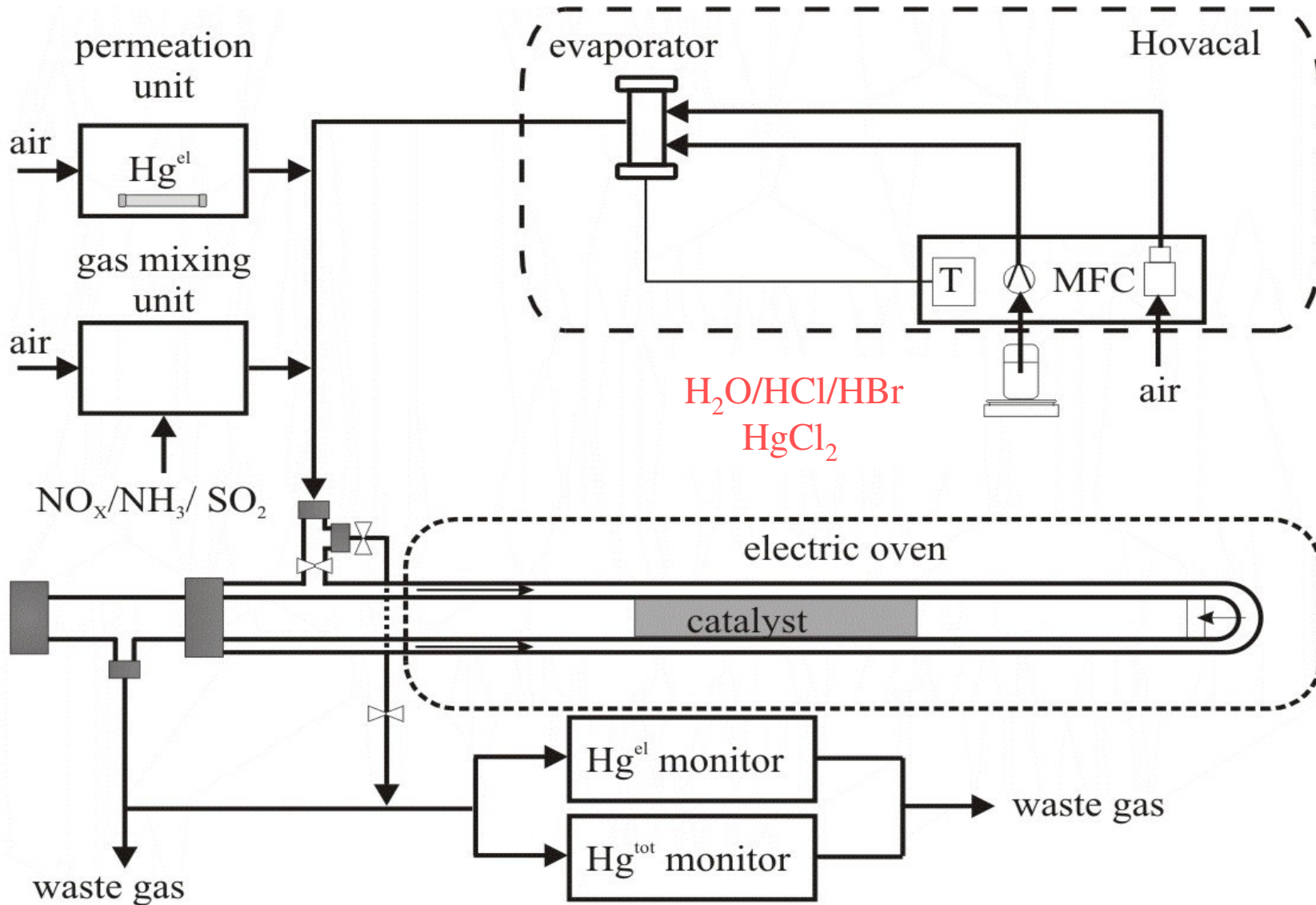


Martin-Luther-Universität
Halle-Wittenberg

TUclausthal

Martin-Luther-University Halle-Wittenberg
Institute for Environmental Engineering
**Heinz Köser, Bernhard Vosteen,
Rico and Ewa Kanefke,
Sandra Straube, Martin Ziegler et al.**

Technical University Clausthal
Institute for Energy Process and Fuel Technology
**Roman Weber, Bernhard Vosteen,
Michael Nolte et al.**

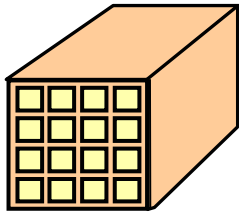


Presented at EUEC 2006:

**Laboratory Research on mercury oxidation
at commercial SCR-DeNO_x catalysts**

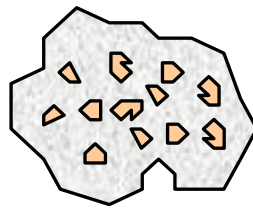
0 - 0,53 - 2,5 - 4,5 Ma.-% V2O5

0 - 2000 µg Hg/Nm³



ca. 15 g ... 31 g

4 x 4 channels
6,5 ... 13 cm long



2 x 0,5 g

crashed



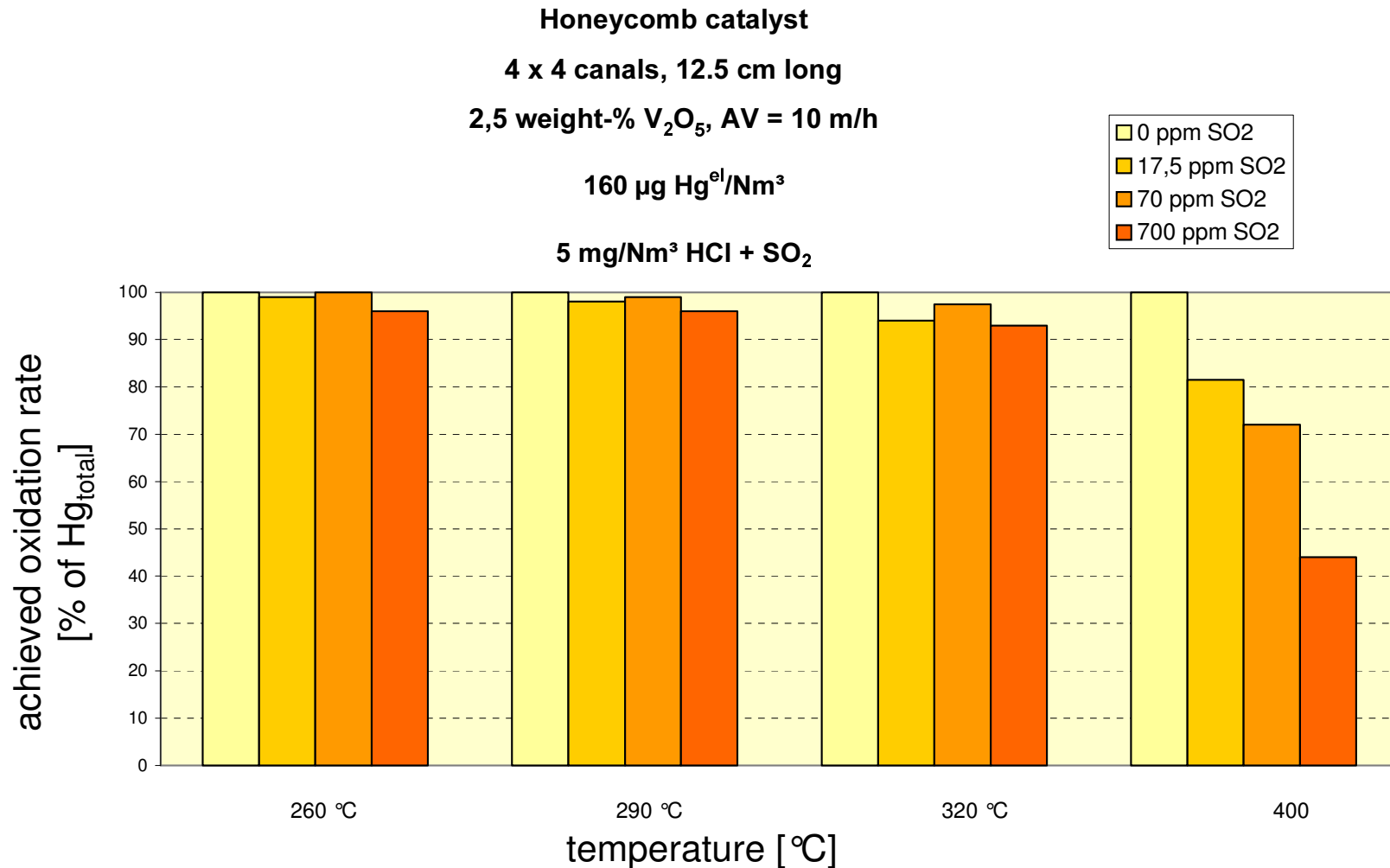
Duration of laboratory test runs:

24 hours up to some days

Duration of laboratory test runs:

Some hours up to one day

post graduate research work by Dipl.-Ing. Sandra Straube



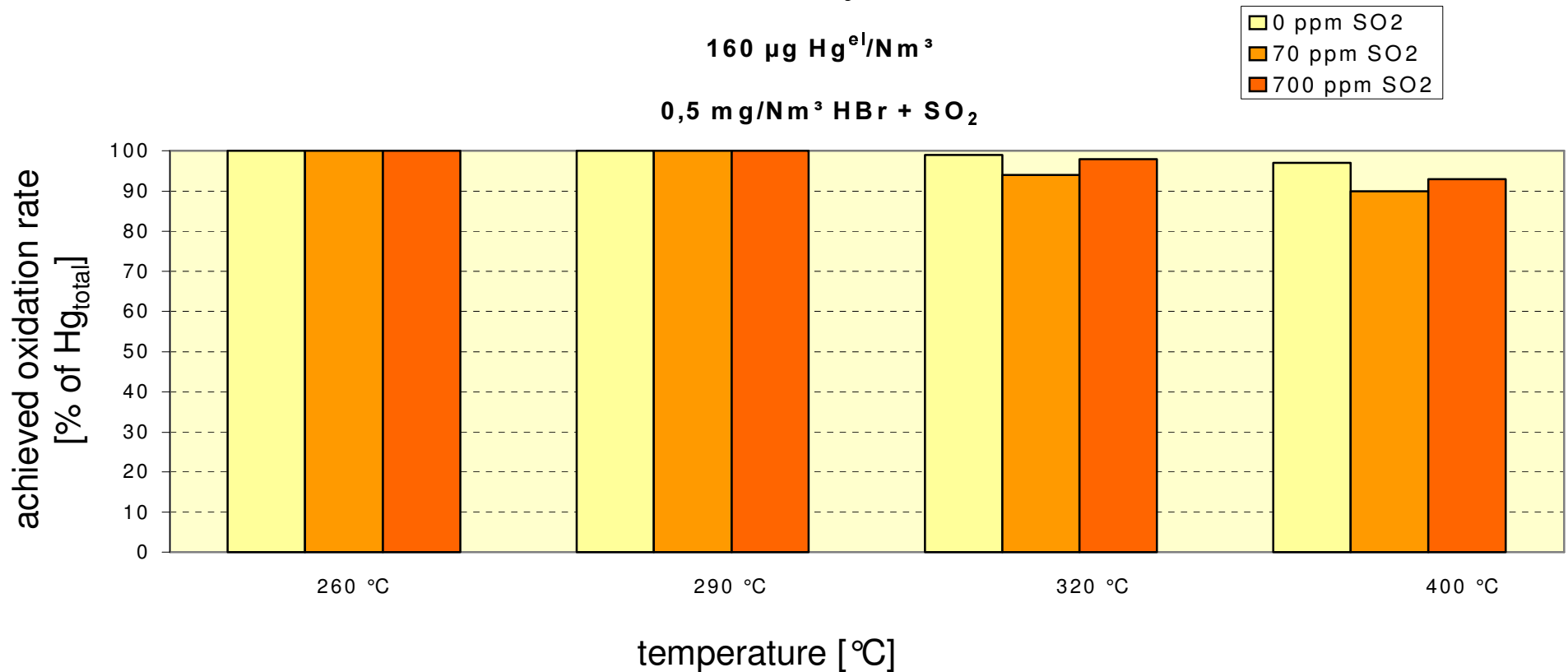
Laboratory tests with humid air, 160 µg Hg^{el}/Nm³ dry and **5 mg HCl/Nm³** added
 (no NH₃ and no NO_x)

Dipl.-Ing. Sandra Straube, MLU Halle-Wittenberg

Influence of Cl₂ and SO₂ on mercury chlorination at SCR catalyst (not denox-active)

Honeycomb catalyst

4 x 4 canals, 12.5 cm long

2,5 weight-% V_2O_5 , AV = 10 m/h160 $\mu\text{g Hg}^{\text{el}}/\text{Nm}^3$ 0,5 mg/Nm^3 HBr + SO_2 

Laboratory tests with humid air, 160 $\mu\text{g Hg}^{\text{el}}/\text{Nm}^3$ dry and **only 0.5 mg HBr/ Nm^3** added
(no NH_3 and no NO_x)

Dipl.-Ing. Sandra Straube, MLU Halle-Wittenberg

Influence of temperature and SO_2 on Hg-bromination at the not denox-active SCR catalyst

In Plant research in 2001

high Hg_{total} in the flue gas (9600 µg Hg/Nm³ dry)

Mass ratio for complete mercury removal

Br/Hg = 500 (“without SCR”)

Laboratory research in 2005

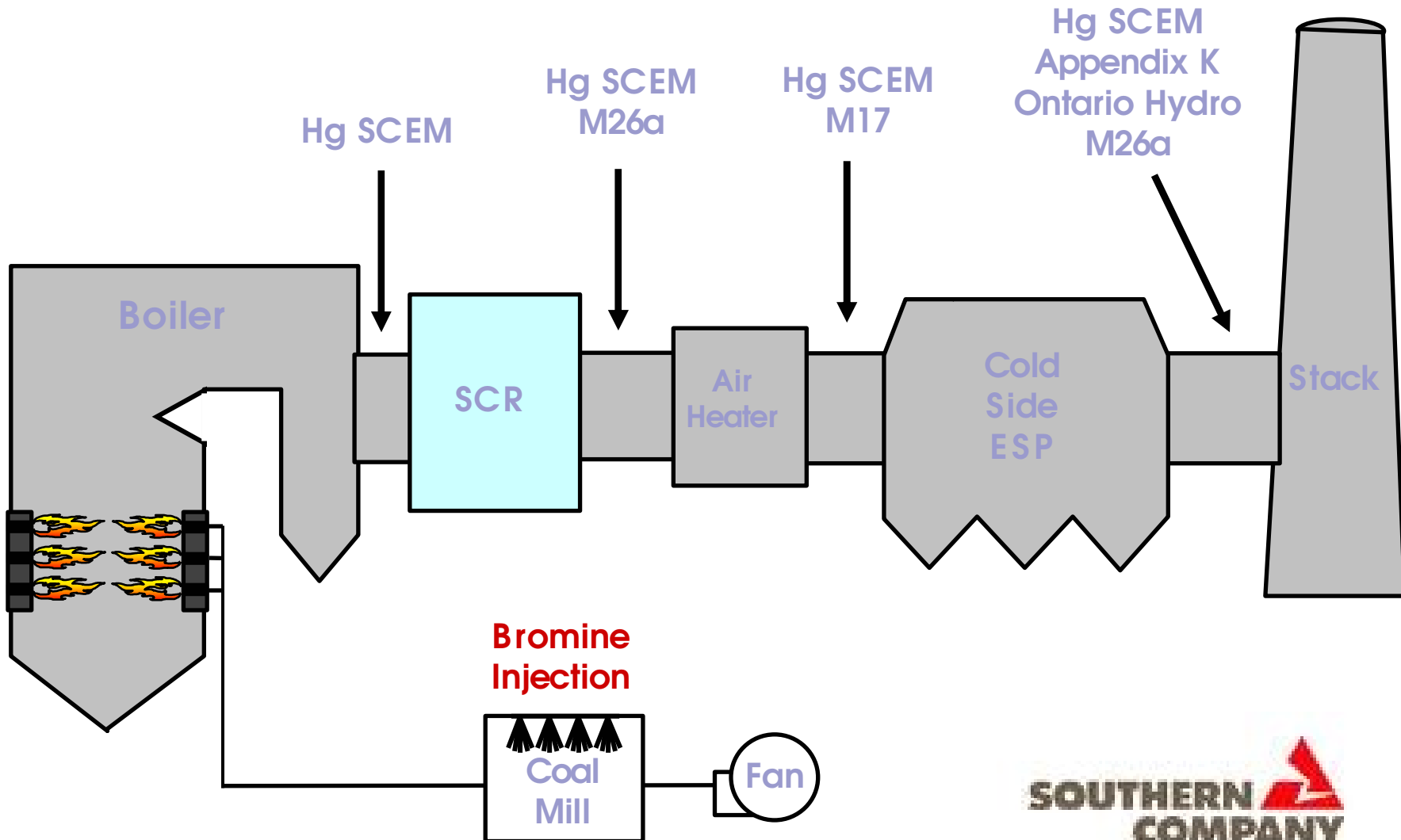
low Hg_{total} in the flue gas (160 µg Hg/Nm³ dry)

Mass ratio for complete mercury removal predicted^{*)}

Br/Hg ≥ ca. 50 (“with SCR”)

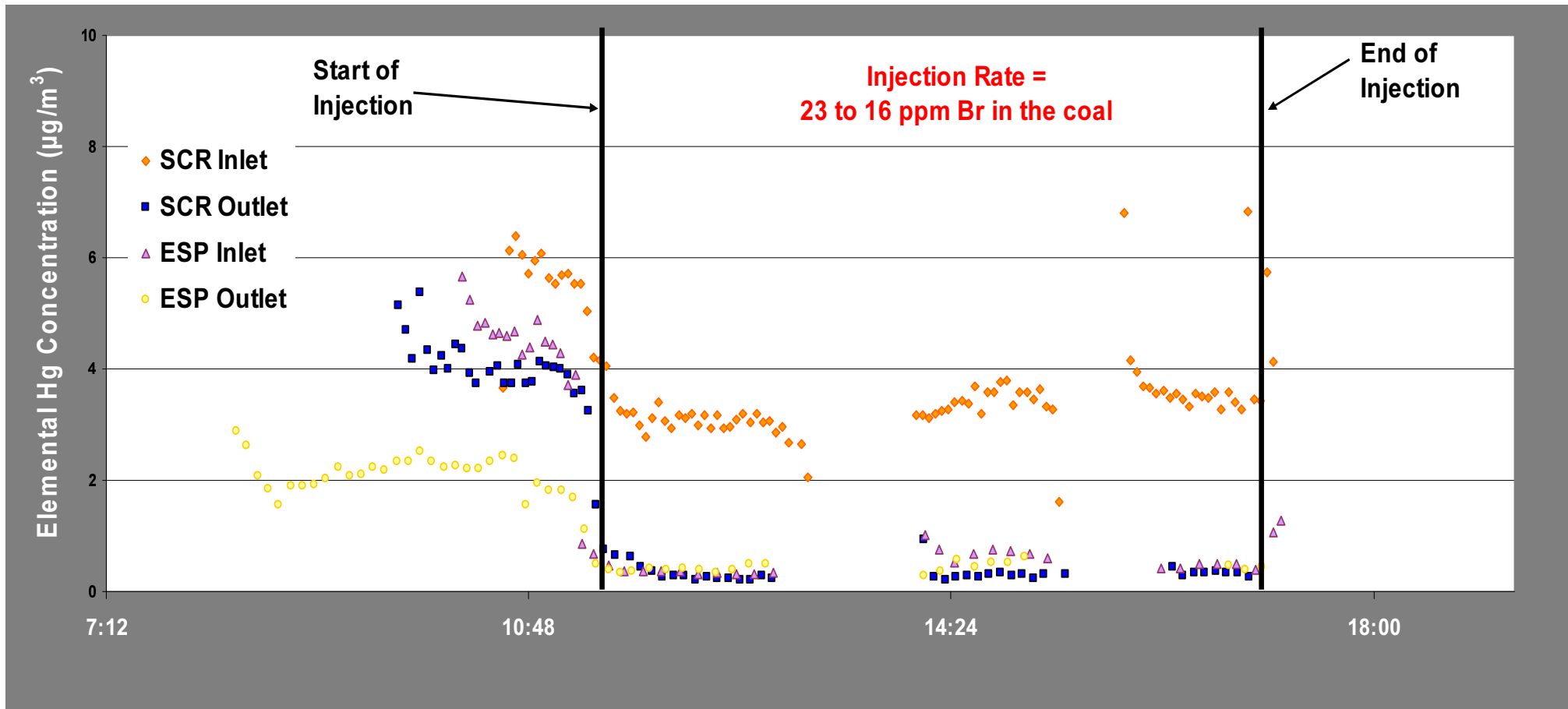
^{*)} Vosteen, B., Straube, S., Koeser, H.: “Mercury Sorption and Mercury Oxidation by Chlorine and Bromine at SCR DeNO_x Catalyst”, Parts A and B, EUEC 2006

Miller Measurement Locations



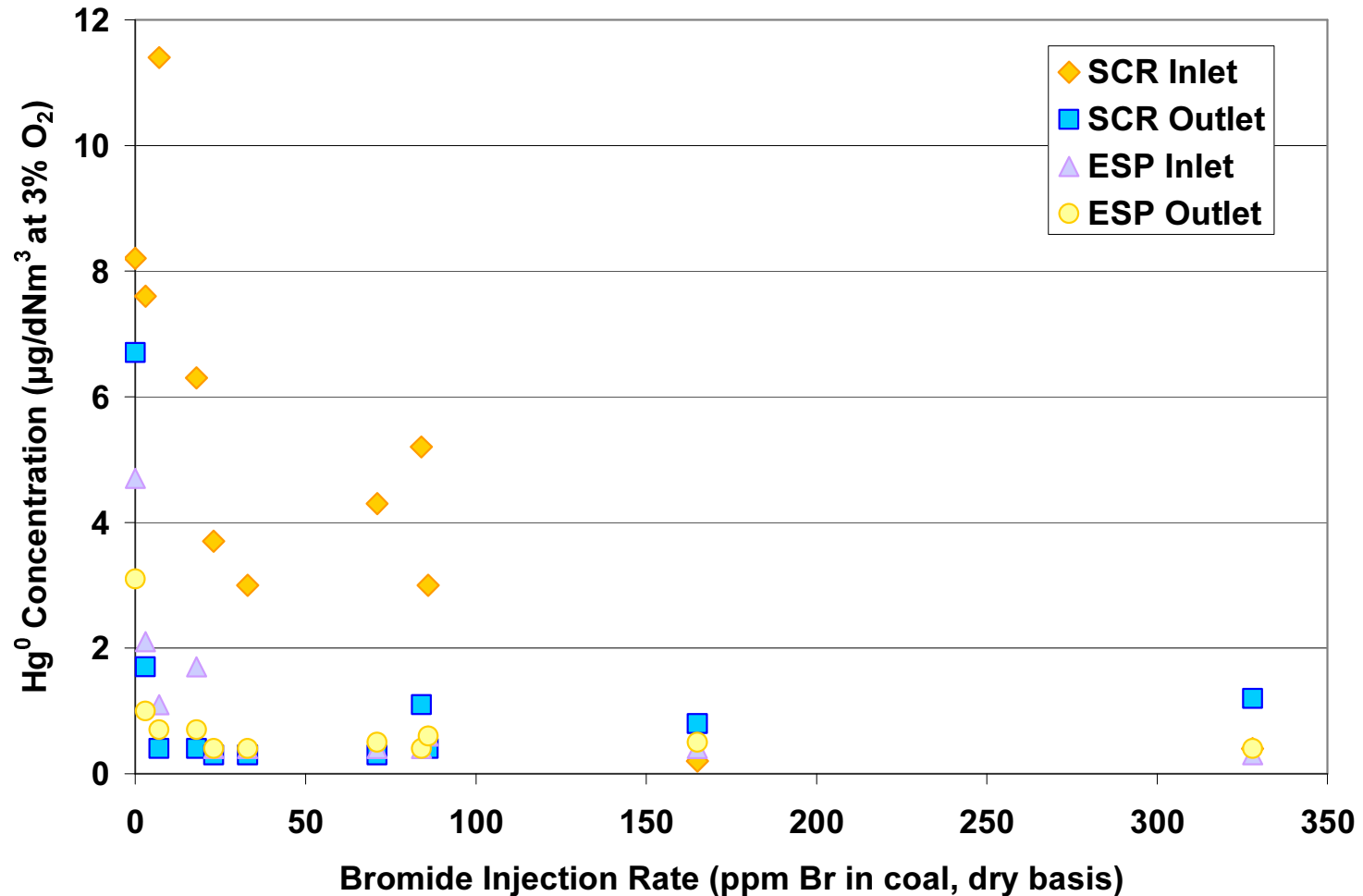
Miller Testing Results

Low Bromine Injection Test



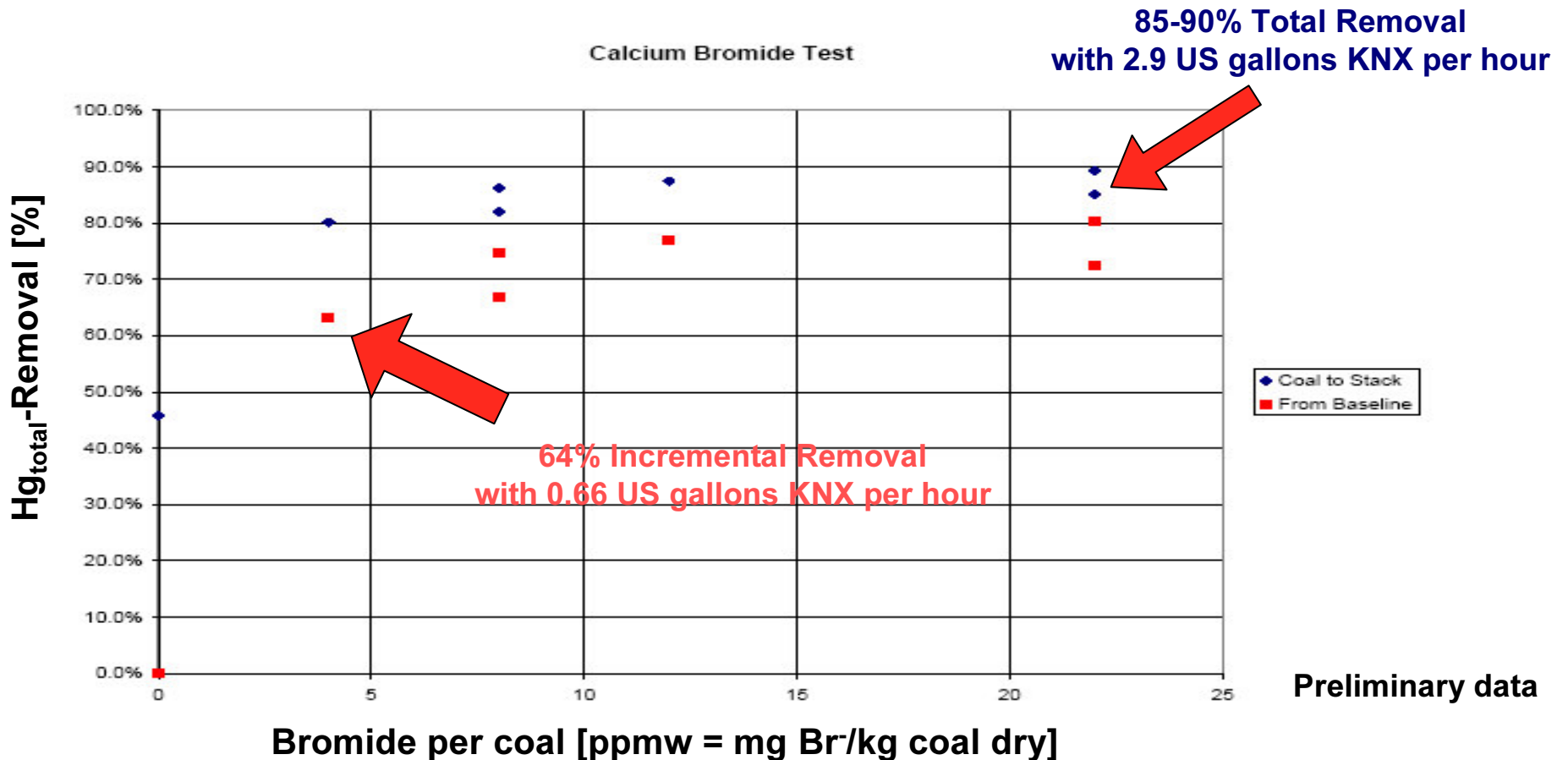
0.5 – 1.0 ppmv Br Concentration

Miller Average Elemental Mercury Concentrations Measured During Each Bromide Injection Test



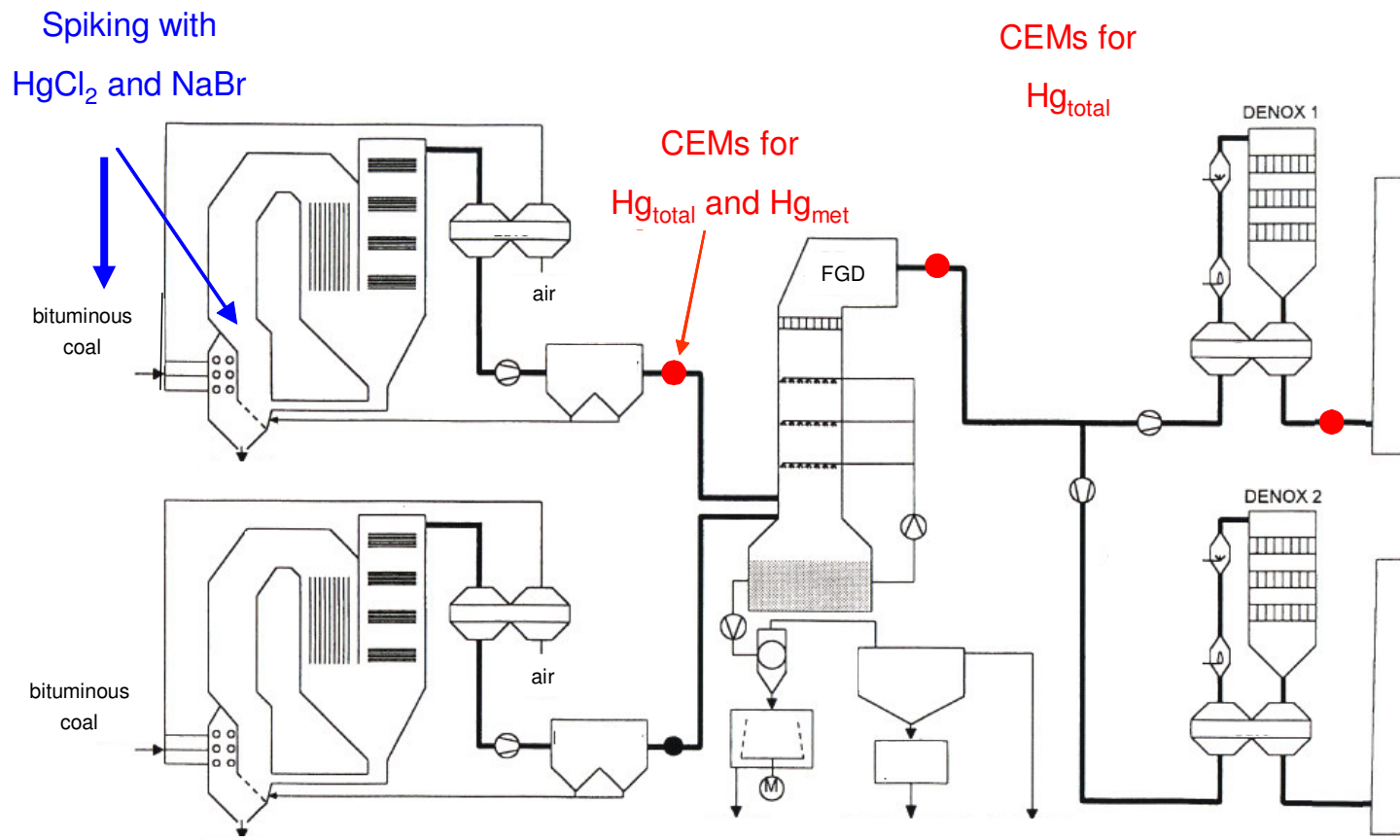
600 MWe, PRB coal, SCR/ESP/WFGD

315 t PRB coal/ h, 0.11 mg Hg/kg coal



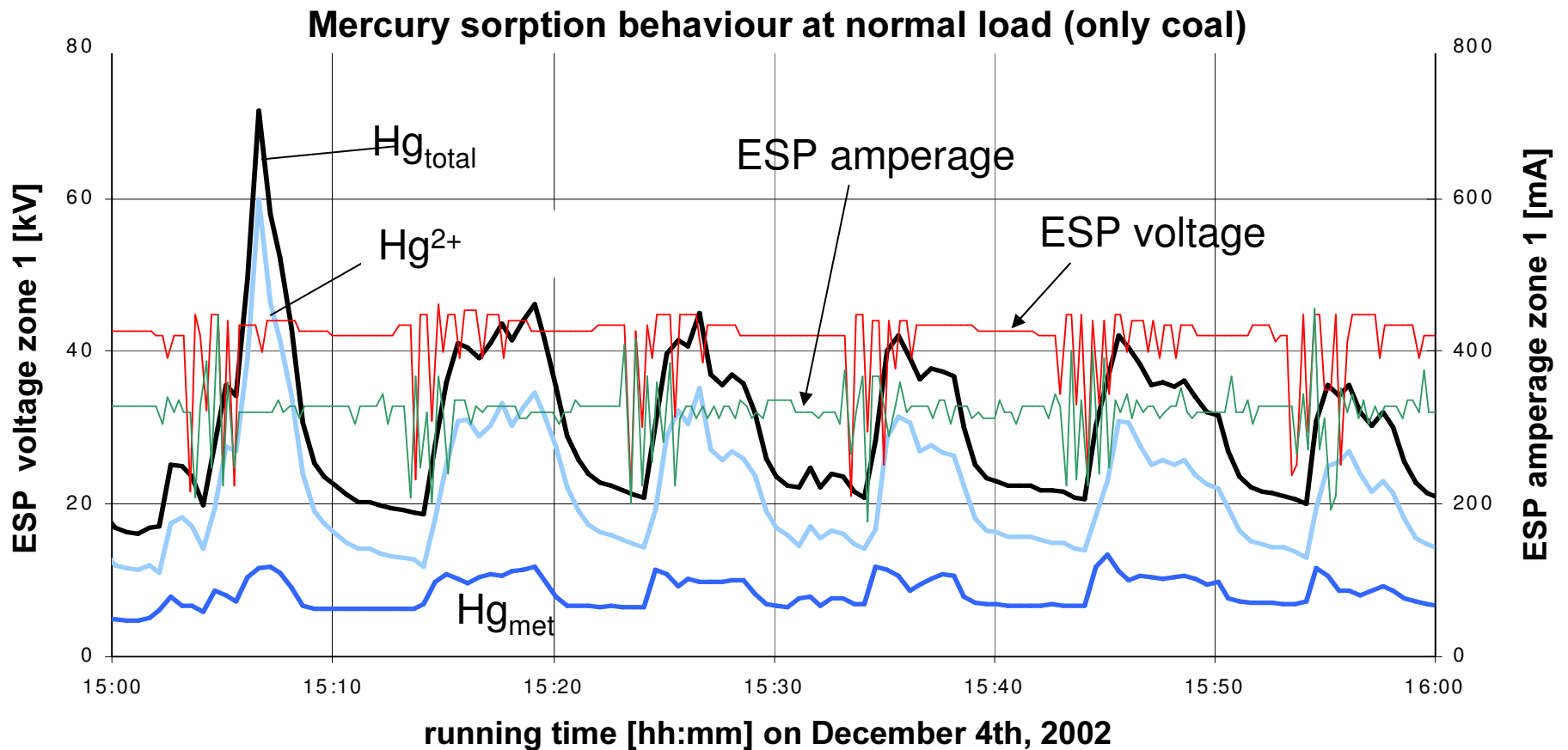
- **First Industrial Demonstration in 2002 (2 x 5 days testing)**
diploma thesis of Michael Nolte

PC-fired Wet Bottom Boiler (100 MW_{therm}; 140,000 Nm³/h dry)
of BAYER AG in Uerdingen (Germany),
ESP, limestone based wet FGD, tail end DeNOx-SCR



diploma thesis of Michael Nolte, 2002

Hg species concentrations in ESP exit gas [$\mu\text{g}/\text{Nm}^3$ dry]

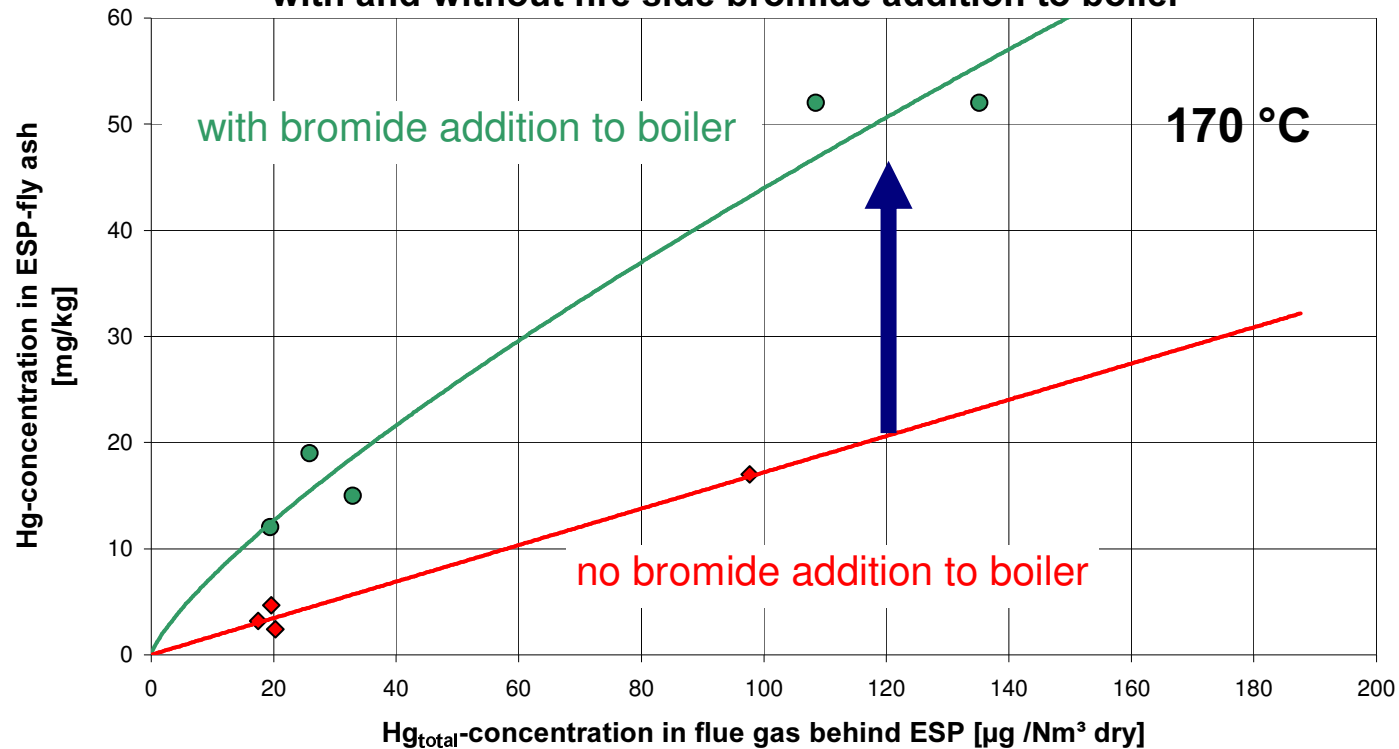


**Mainly ionic mercury is adsorbed at ESP fly ash,
preferably at the fly ash covering the plate electrodes**

diploma thesis of Michael Nolte, 2002

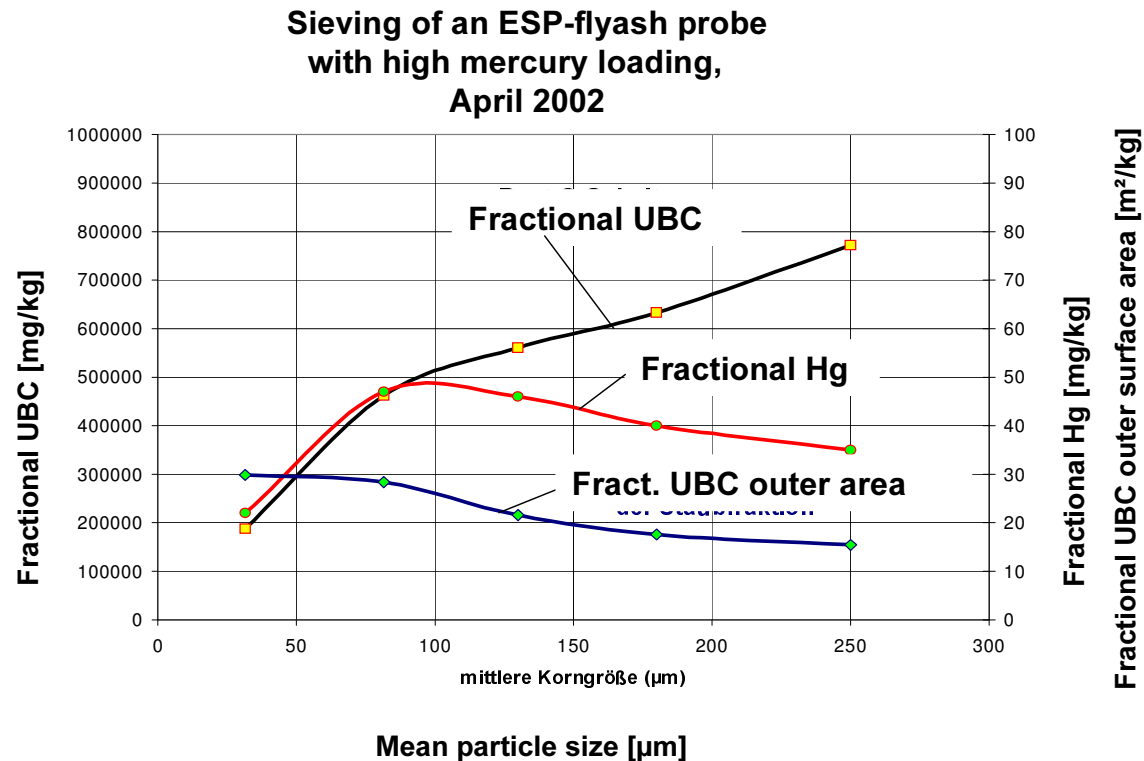
Sorption isothermes of the ESP-fly ash of high LOI (22,3 % UBC)
during test runs in April, 2002

with and without fire side bromide addition to boiler



**Fire side bromide addition (internal bromination)
is improving Hg-adsorption at fly ash UBC by the factor 2,5**

Particle size distribution of UBC and mercury in the ESP-flyash



Conclusions

**„Complete Mercury Oxidation“ achieved by KNX
in Waste Combustion and in Coal Combustion**

- 1. without SCR at a mass ratio Br/Hg > 500**
- 2. with SCR at a mass ratio Br/Hg \geq 50**

*Based on enhanced mercury oxidation,
KNX is improving considerably both
absorptive and adsorptive mercury capture too,
as already demonstrated in 2002*

Thanks for Your attention

Questions?